

BOARD OF INTERMEDIATE EDUCATION, KARACHI
INTERMEDIATE EXAMINATION 2016 (ANNUAL)

Date: 12.05.2016
9:30 a.m. to 9:45 a.m.

CHEMISTRY
(Home Economics Group)

Max. Marks: 07
Time: 15 minutes

The correct answers are
highlighted in red colour.

SECTION 'A'
(MULTIPLE CHOICE QUESTIONS) – (M.C.Qs.)

Code : **HCH-13**

Write this Code. in Answerscript.

NOTE:

- i) This section consists of 14 part questions and all are to be answered. Each question carries ½ mark.
- ii) Do not copy the part question in your answerscript. Write only the answer in full against the proper number of the question and its part.
- iii) The Code of your question paper must be mentioned **in bold letters** in the beginning of your answerscript.

1. Choose the correct answer for each from the given options:

- i) The electronic configuration of Potassium is :
* $1s^2 2s^2 2p^6 3s^2 3p^6 3d^1$ * $1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$
* $1s^2 2s^2 2p^6 3s^2$ * $1s^2 2s^2 2p^6 3s^2 3p^1$
- ii) The product (s) of the reaction between fat and alkali is/are:
* Soap * Detergent * Glycerin * **Soap and Glycerin**
- iii) The common name of Tetrachloromethane is:
* Chloroform * Methyl chloride
* Dimethyl chloride * **Carbon tetrachloride**
- iv) Molecular formula of Propanol is:
* C_3H_8OH * C_3H_5OH * C_3H_6OH * **C_3H_7OH**
- v) 28 gm of KOH equals to:
* **0.5 mole** * 1 mole * 2 mole * 2.5 mole
- vi) This is the main constituent of Vinegar:
* Formic acid * **Acetic acid** * Oxalic acid * Tartaric acid
- vii) The Molecular formula of Octene is:
* C_8H_{18} * **C_8H_{16}** * C_8H_{14} * C_8H_{13}
- viii) Water gas is a mixture of:
* **CO and H_2** * H_2 and O_2 * H_2 and CO_2 * H_2O and CO
- ix) The pH of 0.1M HCl is:
* **1** * 2 * 3 * 4
- x) The maximum number of electrons in M shell is:
* 2 * 8 * **18** * 32
- xi) The number of elements in the third period of periodic table is:
* 2 * **8** * 18 * 32
- xii) This bond is formed by complete transference of electron:
* Covalent bond * Coordinate covalent bond
* **Ionic bond** * Hydrogen bond
- xiii) This is the sweetest sugar:
* Glucose * **Fructose** * Sucrose * Maltose
- xiv) Neutron has:
* positive charge * negative charge
* **no charge** * both negative and positive charges

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