BOARD OF INTERMEDIATE EDUCATION, KARACHI

INTERMEDIATE EXAMINATION, 2016 (ANNUAL)

Date: 04.05.2016 9:30 a.m. to 9:50 a.m.

<u>CHEMISTRY PAPER – II</u>

(Science Pre-Engineering & Pre-Medical Groups)

Tiı	me: 20	minutes	
П	ПΠ	\bigcirc ПП	00

Max. Marks: 17

The correct answers are highlighted in red colour.

SECTION 'A' (MULTIPLE CHOICE QUESTIONS) – (M.C.Qs.)

Code No:GH-09

<u>N(</u>	OTE:	i)		ction consuestion ca			stions an	d all are t	to be ansv	wered.	Write this	Code No.	in the Answerscript.
		ii)	Do not number	Do not copy the part questions in your answerbook. Write only the answer <u>in full</u> against the proper number of the question and its part.									
		iii)	The co	de of your	question	paper is	to be wri	tten <u>in bo</u>	old letters	in the b	eginning	of the ans	swerscript.
1.	Select t	the most appropriate answer for each from the given options:											
	i)	Propana *	l and Propan Chain isome				*		isomers				
		*	Metamers				*	Functio	nal grou	p isomei	3		
	ii)	This is a	natural poly Cellulose	mer:	*	PVC		*	Nylon		*	Terylene	e
	iii)	The nun	nber of mono $2-8$	saccharid *	e units in $\frac{2-10}{}$	oligosac	charides *	is: 2 – 12		*	2 – 20		
	iv)	This is t	he general fo	ormula of	ketones:								
			0			0			0				
		*	$O \parallel R - C - H$		*	R-C	-R	*	R-C	-OH		*	R-O-R
	v)	Another *	name for W Ethyl alcoho		is: *	Methyl	alcohol		*	Propyl	alcohol	*	Butyl alcohol
	vi)	Cod live	er oil is a sou Vitamin A	rce of:	*	Vitamin	В	*	Vitamin	С	*	Vitamin	K
	vii)	This aci	d is used for	etching of	f glass: HCl		*	HBr		*	НІ		
	viii)	Ethyl ac	etate is prese Pineapple	ent in:	*	Orange		*	Guava		*	Lemon	
	ix)	This is u	used to increa Ni	ase the Oc *	tane num Pt	ber and e	efficiency *	of petrol V ₂ O ₅	l:	*	(C2H5)	4Pb	
	x)	-	es of group IV Acidic		Basic		*	Neutral]	*	Ampho	teric	
	xi)	Ruby is	an oxide of: Zinc	*	Alumin	ium	*	Copper		*	Iron		
	xii)	The electronic configuration of the outer shell of an element is $4s^2$, $3d^{10}$, $4p^1$. It belongs to: * IA group and 3^{rd} period * IIA group and 4^{th} period											
		*	IIIA group a				*		oup and	•	od		
	xiii)	H_2S is:											
		*	an oxidizing a sulphonati				*		ing agen ing agent				
	xiv)	Sodium *	amalgam is a	•	f: *	Sodium	and Me	ercury	*	Sodium	and Iron	*	Sodium and Silver

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Al, Cu and Ni

Rhombic sulphur and Monoclinic sulphur are in equilibrium at this temperature:

96.6°C

Rare earth metals

Coinage metals

Al, Cr and Zn

113°C

Fe, Cu and Al

Elements of group IB are called:

Alkaline earth metals

Alkali metals

 $95.5^{\circ}C$

Stainless steel is an alloy of:

Fe, Cr and Ni

xv)

xvi)

xvii)